## Alkenes - AS 2022 Chemistry P2

- 1. June/2022/Paper\_7404/2/No.3
  - 0 3 This question is about isomers.

Hex-2-ene has the molecular formula C<sub>6</sub>H<sub>12</sub>

0 3. 1 Draw the displayed formula of a **position** isomer of hex-2-ene that exists as *E* and *Z* isomers.

[1 mark]

0 3. Draw the displayed formula of a **chain** isomer of hex-2-ene that does **not** exist as *E* and *Z* isomers.

[1 mark]

Butanal has the molecular formula C<sub>4</sub>H<sub>8</sub>O

0 3. Draw the skeletal formula of a **functional group** isomer of butanal that has an absorption in the range 1680–1750 cm<sup>-1</sup> in its infrared spectrum.

[1 mark]

aqasolvedexampapers.co.uk

0 3 . 4	Draw the skeletal formula of a structural isomer of butanal that has an absorption in
	the range 3230–3550 cm <sup>-1</sup> in its infrared spectrum. [1 mark]
0 3 . 5	Several saturated halogenoalkanes contain 17.8% carbon, 3.0% hydrogen and 79.2% bromine by mass.
	Calculate the empirical formula of these compounds.
	Give the IUPAC names of two saturated halogenoalkanes that have this empirical
	formula. [4 marks]
	•
	Empirical formula
	Names of halogenoalkanes
	1
	2

	2.	June/2022/	/Paper	7404/2	/No.10
--	----	------------	--------	--------	--------

Which statement about poly(ethene) is correct?

[1 mark]

A It has a lower relative molecular mass than ethene.

- 0
- B It has a lower density than ethene at standard temperature and pressure.
- 0

C It has a higher melting point than ethene.

0

D It decolourises bromine water.

0

## 3. June/2022/Paper 7404/2/No.11

A polymer is formed from the monomer CH<sub>2</sub>=CHCN

Which statement is not correct?

[1 mark]

A The monomer is propanenitrile.



B The monomer is unsaturated.



C The polymer is an addition polymer.

0
---

**D** The polymer has the repeating unit  $-\overset{|}{c}-\overset{|}{c}$ 



## **4.** June/2022/Paper\_7404/2/No.12

Which alcohol when dehydrated forms a mixture of alkenes?

[1 mark]

A propan-1-ol



B propan-2-ol



C pentan-1-ol

0

D pentan-2-ol

0