AQA - Organic Synthesis - A2 Chemistry P3

1. June/ 2020/Paper_3/No.22

Which has E-Z isomers?

[1 mark]

A C₂H₂Br₂

0

B C₂H₃Br

0

 \mathbf{C} $C_2H_4Br_2$

0

D C₂H₅Br

0

2. June/ 2020/Paper_3/No.23

Which is the mechanism for this conversion?



[1 mark]

A Addition-elimination

0

B Electrophilic substitution

0

C Free-radical substitution

0

D Nucleophilic substitution

3. June/ 2020/Paper 3/No.24

Which compound decolourises bromine water in the absence of sunlight?

[1 mark]

A CH₃CH₂CH₂Br



В



c _____



D CH₃CH₂CHCH₂



4. June/ 2020/Paper_3/No.26

Which does **not** contain an asymmetric carbon atom?

[1 mark]

A CH₃CH(CH₃)CH₂CH₃



B CH₃CH₂CH(CH₃)CH₂CH₂CH₃



C CH₃CH(OH)CH₂OH



D CH₃CH₂CHClCH₃

0

5. June/ 2020/Paper_3/No.27

Which reaction involves addition-elimination?

[1 mark]

A $(CH_3)_2CHBr + KOH \rightarrow CH_3CH=CH_2 + KBr + H_2O$



 $\textbf{B} \ \ \text{CH}_3\text{COCl} \ + \ \ \text{C}_6\text{H}_5\text{OH} \ \rightarrow \ \ \text{CH}_3\text{COOC}_6\text{H}_5 \ + \ \text{HCl}$



C CH₃CH=CH₂ + Cl₂ \rightarrow CH₃CHClCH₂Cl

0

 $\label{eq:decomposition} \textbf{D} \ \ CH_3CH_2CH_2Br \ + \ \ NaOH \ \rightarrow \ \ CH_3CH_2CH_2OH \ + \ \ NaBr$

6. June/ 2020/Paper_3/No.28

Which compound reacts with hydrogen bromide to give 2-bromo-3-methylbutane as the major product?

[1 mark]

A (CH₃)₂C=CHCH₃

0

B CH₃(CH₂)₂CH=CH₂

0

C CH₃CH₂C(CH₃)=CH₂

0

D (CH₃)₂CHCH=CH₂

0

7. June/ 2019/Paper_3/No.21

A possible synthesis of a compound found in jasmine flower oil is shown.

Which mechanism is not used in this synthesis?

[1 mark]

A Electrophilic substitution

0

B Nucleophilic substitution

0

C Free-radical substitution

0

D Nucleophilic addition-elimination

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Three reagents are added separately to four organic compounds.

Which row shows the correct observations?

[1 mark]

| | | Sodium hydrogen carbonate | Acidified potassium dichromate(VI) | Tollens' reagent | |
|---|----------------|---------------------------|------------------------------------|-------------------------|---|
| A | Propan-1-ol | effervescence | orange solution turns green | no visible change | 0 |
| В | Propanal | no visible change | orange solution turns green | silver mirror | 0 |
| С | Propanone | no visible change | no visible change | silver mirror | 0 |
| D | Propanoic acid | effervescence | no visible change | silver mirror | 0 |

9. June/ 2019/Paper_3/No.32

In the UK industrial ethanol is now produced by the direct hydration of ethene. This process has largely replaced the fermentation method.

Which is a likely reason for this change of method?

[1 mark]

| Α | The direct hydration route produces purer ethanol. | 0 |
|---|-------------------------------------------------------------------|---|
| В | The direct hydration route employs milder conditions. | 0 |
| С | The direct hydration route does NOT use a catalyst. | 0 |
| D | The direct hydration route produces ethanol by a slower reaction. | 0 |

| 10. June/ 2019/Paper 3/No.33 | | | | | | | |
|-------------------------------------|------|------|------|--------|----|-----|----|
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Which alkene reacts with hydrogen bromide to give 2-bromo-3-methylbutane as the major product?

[1 mark]

A (CH₃)₂C=CHCH₃

0

B CH₃CH₂CH=CHCH₃

0

C CH₃CH₂C(CH₃)=CH₂

0

D (CH₃)₂CHCH=CH₂



11. June/ 2019/Paper_3/No.34

Which compound can be purified by forming a hot aqueous solution that recrystallises on cooling?

[1 mark]

A Cyclohexene



B Ethanoic acid

C Phenylamine

0

D Benzoic acid