

**AQA – Nuclear Magnetic resonance spectroscopy – A2 Chemistry P3**

1. June/2021/Paper\_3/No.2

0 2

The protein fibroin can be broken down into amino acids using an enzyme.

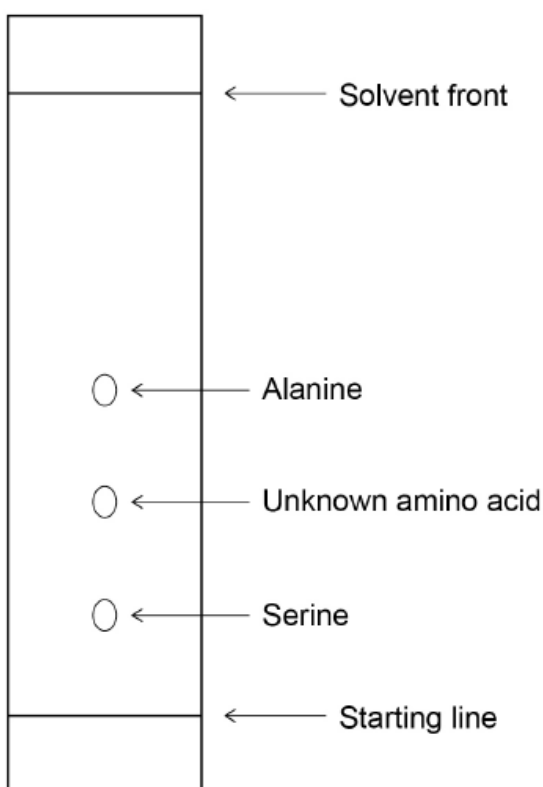
0 2 . 1

A student uses thin-layer chromatography (TLC) to identify these amino acids.

The student identifies two of the amino acids as alanine and serine.

Use **Figure 3** to calculate the  $R_f$  value of the unknown amino acid.

Show your working.

Use your  $R_f$  value and **Table 1** to identify the unknown amino acid.**[2 marks]****Figure 3****Table 1**

Amino acid	$R_f$ value
tyrosine	0.25
glycine	0.34
valine	0.64
leucine	0.73

 $R_f$  value \_\_\_\_\_

Identity \_\_\_\_\_

0 2 . 2 The amino acids cannot be seen as they move during the experiment.

State how the amino acids can be made visible at the end of the experiment.

[1 mark]

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0 2 . 3 State why each amino acid has a different  $R_f$  value.

[1 mark]

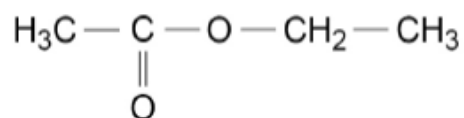
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2. June/2021/Paper\_3/No.11

Which statement does **not** support the suggestion that an unknown organic compound is



[1 mark]

- A Its  $^1\text{H}$  NMR spectrum has 3 peaks with an integration ratio of 2:3:3
- B Its  $^{13}\text{C}$  NMR spectrum has 3 peaks.
- C Its infrared spectrum has an absorption at  $1735\text{ cm}^{-1}$
- D It has 36.36% by mass of oxygen and 9.09% by mass of hydrogen.