

AQA – Amines – A2 Chemistry P2

1. June/ 2019/Paper_2/No.1

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This question is about amines.

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Give an equation for the preparation of 1,6-diaminohexane by the reaction of 1,6-dibromohexane with an excess of ammonia.

[2 marks]

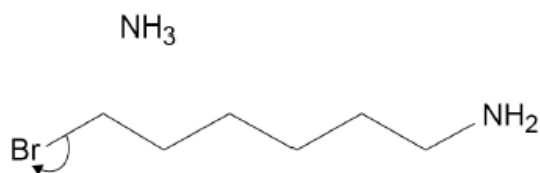
0 1 . 2

Complete the mechanism for the reaction of ammonia with 6-bromohexylamine to form 1,6-diaminohexane.

Suggest the structure of a cyclic secondary amine that can be formed as a by-product in this reaction.

[4 marks]

Mechanism



Cyclic secondary amine

- 0 1 . 3** 1,6-Diaminohexane can also be formed in a two-stage synthesis starting from 1,4-dibromobutane.
Suggest the reagent and a condition for each stage in this alternative synthesis. **[3 marks]**

Stage 1 reagent and condition

Stage 2 reagent and condition

- 0 1 . 4** Explain why 3-aminopentane is a stronger base than ammonia. **[2 marks]**

- 0 1 . 5** Justify the statement that there are no chiral centres in 3-aminopentane. **[1 mark]**
